The Earliest Steel from Transjordan

Corinthian Bronze: Rome's Purple Sheen Gold

Evidence for a Nineteenth Century Forge at Catoctin, Maryland

Scientific Dating of the San Marco Horses

Folding: A Prehistoric Way of Working Native Copper in the North American Arctic

Analysis of Two Bronzes from a Nigerian Asunaja Shrine

The Prehistoric Copper Smelting Industry at Cerro de los Cementerios, Peru: Analysis of the Product

An Early Steel Implement from Baradih, Bihar Province, India: Metallurgical Studies

Lead Isotope Analyses of Late Period Egyptian Bronzes

Book Reviews

This special MASCA Journal supplement is issued to coincide with the Science Fair which will be devoted to archaeometallurgy, at the annual meeting of the Archaeological Institute of America being held in Philadelphia, December 27th - 30th. The supplement's articles themselves underline two facts. First, that archaeometallurgy has developed as a major element of MASCA's research activity over the past three years, to cover the metalworking technology of many cultures in a wide matrix of time and space. Second, that archaeometallurgy is also a rapidly growing discipline worldwide.

I am delighted therefore to take this opportunity to thank those outside the University of Pennsylvania for their tireless support of the MASCA program—specifically Michael Notis and Heidi Moyer at Lehigh University, Charles Swamy at the University of Delaware, and Reed Knox (now retired from Alan Wood Steel)—and to express my appreciation for the excellent contributions provided by our colleagues abroad, contributions which make this supplement thematically well-rounded overall.

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Plate 1: Ankle/Bracelet types from Cave A4 (D, approx. 0.07 m, 0.08 m, respectively). (Photographs courtesy of N. Hartmann, MASCA.)
plastic, polished on four abrasive papers and finally on three polishing wheels. All samples were studied in both the mounted and etched condition and photographed. The etchant used was 2% nitric. When slag inclusions were found to be present, their chemistry was studied using energy dispersive x-ray analysis in a scanning electron microscope.

A preliminary analysis of five artifacts, the only specimens found to have macroscopic islands of uncorroded metal, is presented here. A total of seventeen artifacts in all were studied. The remaining twelve artifacts were found to be heavily oxidized and are now being studied for evidence of pseudomorphic ghost structures indicative of carburization (see Knox 1963).

Table 1 provides the elemental analyses of the five artifacts with uncorroded metal as obtained by PIXE analysis (Folkman 1973, Fleming and Crowfoot-Payne 1979). In terms of impurities present we note that those elements which might have affected the composition and functional behavior of the metal, such as phosphorus which can harden or embrittle the metal, occur at minimal concentrations. Additionally the small amounts of calcium and aluminium point to the ore source being a high purity one, and that no slagging additions were made to the smelt.

**Metallurgical interpretations**

1.55: The extent of carburization of the metal of this artifact is illustrated in Plate 2. The coarse-grained microstructure varies from fully pearlitic to pearlitic with ferrite appearing at the prior grain boundaries. The proportions of pearlite in the microstructure are consistent from field to field, and at higher magnification the pearlite appears to be spheroidized.

Widmanstätten ferrite plates were observed within the prior grain boundaries, in elongated angular structures running parallel to one another (Plate 2). However Widmanstätten patterning was also found within the grains as well, an unusual feature probably owing its stability or formation to the co-occurrence of nitride needles in the ferrite phase (Plate 3). This structure is indicative of long-time exposure in the smelting furnace and/or the smith's forge, as well as of subsequent slow cooling. Additionally the SEM analysis of the fayalite (Fe$_2$SiO$_4$) slag-matrix of slag particles entrapped in the metal indicated that they contained a relatively high level of potassium, which would point to a charcoal fueled process, and only a small volume fraction of wustite (FeO) indicative of the high efficiency and skill of the smelting operation.

Plate 2: 1.55 SEM photomicrograph (3000x) showing nitride needles in ferrite surrounded by partially spheroidized pearlite.

1.77: One surface of this bracelet had a relatively high carbon content and a fine ferrite grain size, but the carbon content was still short of the eutectoid composition which would be about 0.7% carbon. The remainder of the cross-section revealed only minimal evidence of carburization distributed unevenly among large grains of ferrite.

Elongated slag stringers record the flattening which the metal underwent during forging (Plate 4). Working was extensive enough that certain stringers were themselves broken into smaller sets of aligned slag inclusions, with cracks running transverse to the direction of elongation. Clearly the forging was carried out at a lower temperature, I.e., while the slag was brittle, not plastic.

SEM analysis of slag particles in this artifact indicated that they comprised a typical wustite-fayalite composition with a relatively high volumetric contribution of wustite. Also, only very low levels of sodium and potassium were found in the slag. These observations, when contrasted with the results for 1.55, could indicate a highly variable smelting process, or a completely separate source of smelted iron.

Table 1

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<th>Sample reference**</th>
<th>Fe</th>
<th>Al</th>
<th>Si</th>
<th>S</th>
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<td>0.010</td>
<td>0.039</td>
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<td>0.010</td>
<td>0.010</td>
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</tr>
<tr>
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<td>0.017</td>
<td>0.10</td>
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*Elements also sought, but not found in concentrations above their detection limits using the PIXE method: K, 0.0038; Mn, 0.0008; Zn, 0.0008; and Ni, 0.0058.
**a and b designations denote non-contiguous islands of intact metal in the same mounted sample.
I.147: The metal of this artifact is a coarse-grained pearlite steel (Plate 5), with a carbon content of about 0.7%, perhaps a little less because of the presence of a small amount of ferrite. There is evidence of partial spheroidization of the carbides, and Widmanstätten side plates had formed in hypoeutectoid regions along one edge of the bracelet. Slag inclusions appear to be mostly lathite and of a single phase.

Plate 6

I.147 Photomicrograph (400X) of coarse-grained pearlite steel with partially spheroidized carbides and angular Widmanstätten patterning.

I.226: The microstructure of this artifact varies from large-grain ferrite to ferrite plus pearlite (Plate 6). The composition is well below that of the eutectoid and what carbonization is present is quite irregular in distribution. Subgrain boundaries were detected within single grains of ferrite, and are presumably a result of lower temperature working. The slag chemistry of this artifact is similar to that of I.177 discussed above.

I.302: This artifact contained hypoeutectoid steel (approximately 0.85%C) with large, coarse carbides evenly distributed through the metal, and present even in the oxide (see Plate 7). They, like the pearlite present, were spheroidized. In this instance there were virtually no inclusions upon which to base a discussion of the slag chemistry related to production of this artifact.

Plate 7

I.203 Photomicrograph (400X) of large coarse spheroidized carbides evenly distributed in the hypoeutectoid steel.

Preliminary conclusions

The initial results of this study of the five anklets/bracelets are both exciting and enigmatic. Four of the artifacts have a structure wherein there is a uniform distribution of carbon from surface to surface, making them the earliest verified instances of mild steel from Jordan and placing them with a small group of the earliest dated steel from Eastern Mediterranean sites such as Edom on Cyprus (Tholander 1971) and Tzur nach in Palestine (Steck-Wheeler et al. 1981). Initial investigation of an additional twelve oxidized iron artifacts from Cave A4 suggests that the majority of these may also have been composed of a mild steel.

Why pieces of jewelry should be manufactured in steel is not clear, nor is the process by which this steel was produced clearly definable. The definitive evidence for solid state carbonization would be the presence of a carbon gradient at or near the surface, i.e., carbon diffusion into the iron that would have proceeded from the exterior surface inwards. The metallography of the Ba'ajah artifacts however, revealed no gradients of this kind in the cross-sections taken, in part because of surface corrosion, in part because the microstructure of the metal is so uniform. Thus it is not possible to draw a definitive conclusion as to the origin of the material via carbonization.

An alternative possibility can be suggested which would account for the microstructures evident in the Ba'ajah steel. In most of the artifacts not only are the amounts of carbon near the eutectoid composition (0.7%) but also there are high concentrations of nitride needles present in the ferrite (Plate 5). This could be explained if the iron being reduced had reached a semi-molten (slushy) or molten state in the smelting furnace. Though the temperatures required to achieve this state, 1400-1475°C, are well above the normal range thought to have been routinely used in antiquity, they are not actually beyond the capacity of a bellows-blown furnace. Localized regions of the bloom, or the entire bloom itself, could therefore, under some unusual yet conceivable conditions have been melted molten (see Smith 1969), yielding appropriate high diffusion rates and high solubilities of both nitrogen and carbon into the liquid iron.

However, though the Ba'ajah steel artifacts show no indication of having been subjected to any heat treatment such as quenching or tempering, this metal would have had mechanical properties (i.e., strength) which would have made it at least equivalent to the bronzes (with about 10% to 15% tin content) found in the same tomb.

In Cave A4, bronze artifacts were three times as prevalent as steel ones, yet intriguingly, none of these artifacts were weapons. The available archaeological and documentary evidence for the Late Bronze/Early Iron Age transition in Palestine suggests that it was far from peaceful. It is during this period that the Israelites and Ammonites begin to consolidate their empires in the region.

We have no reason to believe the iron artifacts were imported. The proximity of iron ore deposits of limonite and hematite in the Wadi Ram and Ahlan regions, the nearest one being only 10 kilometers to the north (Basha 1969), Bender 1974, together with the fact that virtually nothing else recovered from Cave A4 apart from perforated Mediterranean and Red Sea mollusk species could be considered as imports, strongly suggests nearby production, perpetuating Bronze-Age metalworking traditions.

Limited archaeological investigation in the Ajlun at the uits of Moghazar et al. Wardib and Abu Thawab (Coughon 1978) has thus far only been able to substantiate medieval Islamic smithing operations, but surface collection of sherds from the Roman/BYZantine and early Iron periods points to earlier activity. However, until these areas, including the Ba'ajah, are more fully surveyed and well-dated metal-working installations and associated settlements found (e.g. Khirbet Umm ad-Dandarim see McGovern 1980), such a hypothesis, or alternative views which see the introduction of iron/steel in Transjordan as more abrupt and a result of an intrusive cultural element, remain unproved.

Acknowledgments

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